



**SYNTHESIS AND ANTI-TUBERCULAR ACTIVITY- ;  
N,N'-BIS(BENZOTHIOZOLYL)-2,6-  
PYRIDINEDICARBOXAMIDE**

Suparna Sharma<sup>\*1</sup>, Anuj Kumar Sharma<sup>1</sup>, Ompal Singh<sup>1</sup>, Umesh  
Kumar Singh<sup>1</sup>, Ashish Kumar Sharma<sup>2</sup>

1.Kharvel Subharti College of Pharmacy, S.V. Subharti University, Meerut  
2.Department of Chemistry, IIMT College of Medical Sciences, Meerut.

**Abstract**

N,N'-bis(benzothiazolyl)-2,6-pyridinedicarboxamide was synthesized by the condensation of 2,6-Pyridinedicarboxylic acid and thionyl chloride under anhydrous conditions, with benzothiazole, synthesized compound was characterized by IR (KBr)  $\nu$   $\text{cm}^{-1}$ ,  $^1\text{H-NMR}$  (DMSO- $d_6$ ),  $^{13}\text{C-NMR}$  and mass spectral studies. The results show that the substitution of 2-amino-benzothiazolyl group at -OH moiety of carboxylic group of pyridinedicarboxylic acid makes this molecule active against Mycobacterium -tuberculosis.

**Keywords:** - N,N'-bis(benzothiazolyl)-2,6-pyridinedicarboxamide, M-tuberculosis

**Introduction**

The amide bond -C(O)NH- of Pyridine-2,6-dicarboxamide has long attracted much attention since it is an essential building unit in proteins. The high stability of the amide linkage towards hydrolysis is of crucial importance to biological systems, since it allows the construction of peptides from relatively simple amino acid precursors[1].

Another derivatives of pyridine- dicarboxamide in which pyridine ring is in the form of 1,4-dihydropyridine are used as anti-allergic[2-4] and anti-inflammatory agents[5-7]. They are also used in the treatment of cardio-vascular diseases[8] as well as potential cardiovascular agents[9-10]. In recent years Pyridine-2,6-dicarboxamide derivatives are used in asymmetric synthesis, molecular receptor[11-12], & in dendimer synthesis[13]. Pyridine-2,6-dicarboxamide derivatives show their various biological activities. N,N'-2-naphthyl-Pyridine-2,6-dicarboxamide exhibit its sedative[14] and anticonvulsants properties [15]. Another derivatives of pyridine dicarboxamide in which

Nitrogen atom of pyridine dicarboxamide is substituted with heterocyclic moiety like isoxazoles are used as anti-allergic[16] and anti-inflammatory agents [17]. Sulphonamides substituted pyridine dicarboxamide are also used in the treatment of cardiovascular diseases[18] as well as they are potential cardiovascular agents[19]. Our ongoing research is to synthesize the benzothiazolyl substituted pyridine dicarboxamide derivatives and evaluate their in-vitro activity against M-tuberculosis which are resistance to Moxifloxacin.

**EXPERIMENTAL**

**MATERIAL AND INSTRUMENTATION**

All the solvents used were of analytical grade. The starting materials, 2,6-pyridinedicarboxylic acid, thionyl chloride and 4-amino-benzothiazole were of BDH grade. The carbon, hydrogen and nitrogen contents of the ligand and the complexes were determined using a Heraeus C H N rapid analyzer. IR Spectra in the 4000-400  $\text{cm}^{-1}$  range were measured on a Thermo Nicolet 320 FTIR spectrometer using KBr discs.  $^1\text{H-NMR}$  spectra were recorded in DMSO- $d_6$  as the solvent at 400 MHz on a BRUKER AMX 400 spectrometer using tetramethylsilanes(TMS) as an internal reference.

**SYNTHESIS OF THE COMPOUND**

2,6-Pyridinedicarboxylic acid (1.67g) and thionyl chloride (20 - 25 ml) were refluxed under anhydrous conditions for 4 - 6 h until a clear solution was

**Correspondence Address:**

Suparna Sharma

Asstt. Professor

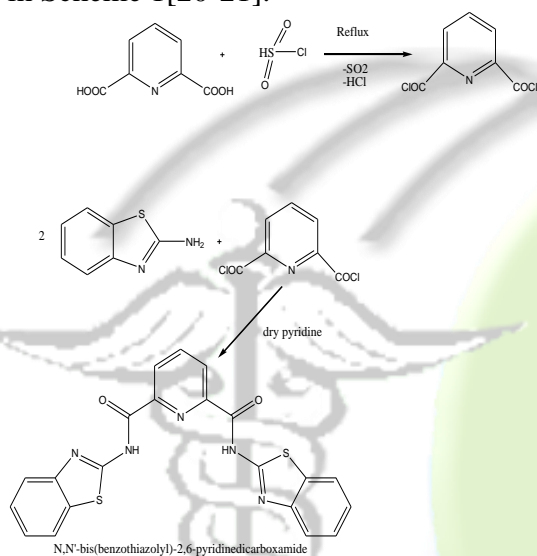
Kharvel Subharti College of Pharmacy

S.V. Subharti University, Meerut 25001

Email: [suparna\\_anuj@rediffmail.com](mailto:suparna_anuj@rediffmail.com)

Mobile no:- +91-9412525845

obtained. The excess thionyl chloride was removed under reduced pressure and the remaining mixture was cooled to 0 °C and dry pyridine (30 – 35 ml) was added followed by 4-amino-benzothiazole (3.0g,) with occasional stirring, until the evolution of HCl had ceased. The solid obtained was poured into ice cold water (200 ml), filtered off and washed with 5 % sodium bicarbonate solution, then with hot water and ethanol. The resulting compound was recrystallized from dioxane. The purity of the compound was controlled by TLC on pre-coated silica gel, which gave a single spot. The synthetic route of *N,N'*-bis(benzothiazolyl)-2,6-pyridinedicarboxamide (BPD) is given in Scheme 1[20-21].



### Microanalysis

The elemental studies, melting point studies and solubility of synthesized compounds shown by table 1

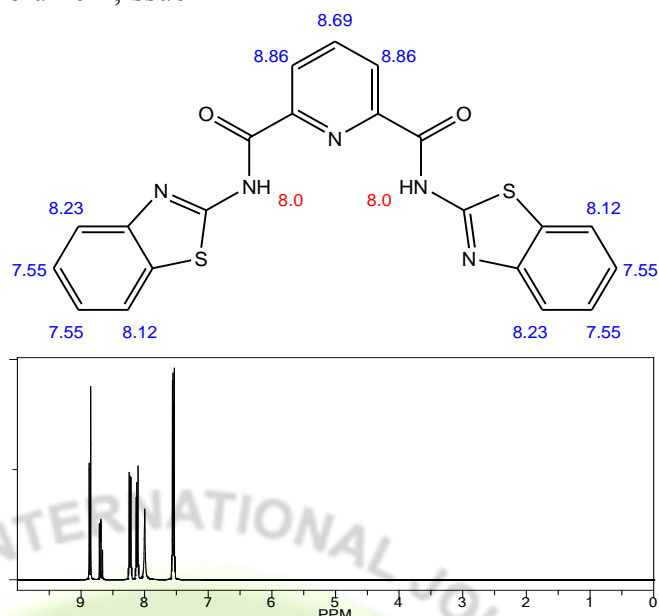
**TABLE-1; Elemental analysis of synthesized compound**

Compound	C% Calc./found	H% Calc./found	N % Calc./found	Solubility	MP°C
C <sub>22</sub> H <sub>15</sub> N <sub>4</sub> S <sub>2</sub> O <sub>2</sub>	58.45/58.46	3.03/3.02	16.23/16.27	DMSO	175

### SPECTRAL STUDIES

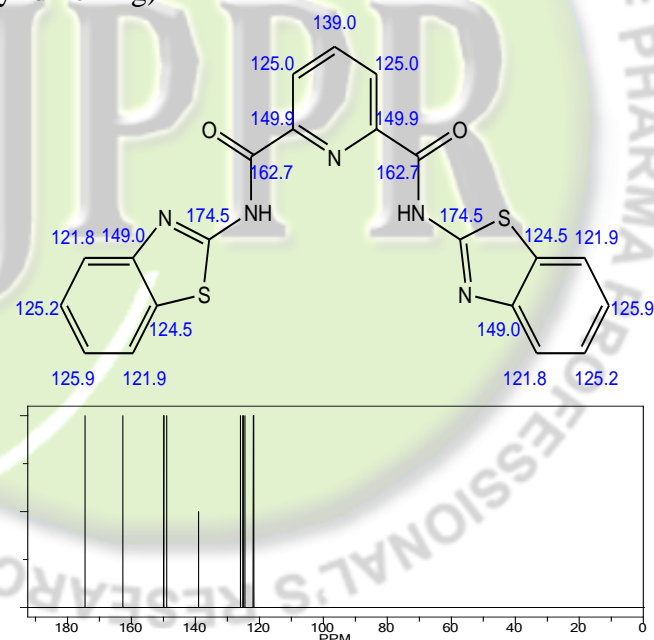
**IR (KBr)  $\nu$  cm<sup>-1</sup>** 3252m (NH), 1696s (Amide-I), 1604s (Benzothiazole), 1610m 1542m 1456m(Pyridine ring stretching ),1468(Amide II) ,998 w (Pyridine ring Breathing)

**<sup>1</sup>H NMR(DMSO-d<sub>6</sub>)** N1H and N2H 13.50 (s, 2H) H3 and H5 8.53 (d, 2H; J = 10.4 Hz) H4 8.25 (t, 1H; J = 8.0 Hz) H8' and H8'' 7.92 (d, 2H; J = 11.2 Hz) H5' and H5'' 7.82 (d, 2H; J = 10.4 Hz) H7' and H7'' 7.49 (t, 2H; J = 8.80 Hz) H6' and H6'' 7.36 (t, 2H; J = 8.0 Hz)



**<sup>1</sup>H NMR spectra of *N,N'*-bis(benzothiazolyl)-2,6-pyridinedicarboxamide**

**<sup>13</sup>C NMR(DMSO-d<sub>6</sub>)** 174.5(carbon atom of thiazole ring),124.4,121.9 , 125.9,125.2,121.8,149.0 (Aromatic carbon atom)139,125,149,149.9,125,(Carbon atom of pyridine ring)



**<sup>13</sup>C NMR Spectra of *N,N'*-bis(benzothiazolyl)-2,6-pyridinedicarboxamide**

**MS:m/z(%) (M<sup>+</sup>)** m/z = 638,

### BACTERIAL CULTURE

The strain of bacteria used was MDR resistant Mycobacterium –Tuberculosis. All strains were isolated from L.L.R.M. Medical college, Meerut. The identity of all the strains was confirmed. A bacterial pension was prepared and added to the sterilized medium before solidification .The media with bacteria was poured in to sterilized Petri dishes under aseptic conditions. Different weights of synthesized compounds

(6.250µg/ml, 3.125µg/ml, 1.562µg/ml ) were placed in the surface of culture and incubated at 37°C for 24hours. After incubation the zone of inhibition (mm) recorded in table 2

**Table -2 ANTI-TUBERCULAR ACTIVITIES OF SYNTHESISED COMPOUND**

Compound	Concentration I	Concentration II	Concentration III
C <sub>22</sub> H <sub>15</sub> N <sub>4</sub> S <sub>2</sub> O <sub>2</sub>	1.562 µg/ml	3.125µg/ml	6.250µg/ml
Zone of inhibition(mm)	9mm	12-15 mm	22-23mm

Moxifloxacin = Z.I.= 23-25mm

## RESULT AND DISCUSSION

The synthesized compound show its maximum activity against M-tuberculosis in 6.25µg/ml concentration in D.M.S.O. At this concentration reported compound show 22-23 mm zone of inhibition which is quietly similar to zone of inhibition of Moxifloxacin standard drug (ZI= 23-25 mm).

These results show that the substitution of 2-amino-benzothiozoyl group at -OH moiety of carboxylic group of pyridinedicarboxylic acid make this molecule active against Mycobacterium tuberculosis.

## REFERENCES:-

- 1).Eharkar PS, Desai B., Vora V., Three dimensional quantitative structure-activity relationship of 1,4-dihydropyridines as antitubercular agents. *J. Med. Chem.* 2002; 45: 4858-4867
- 2).Tasaka HS, Ohmori H, Synthesis and structure-activity analysis of novel dihydropyridine derivatives to overcome multidrug resistance. *Bioorg. Med. Chem. Lett.* 2001; 11:275-277
- 3).Amin L, Navidpour AS, Synthesis and antitubercular activity of new N,N-diaryl- 4-(4,5-dichloroimidazole-2-yl)-1,4-dihydro-2,6-dimethyl-3,5 pyridinedicarboxamides. *DARU.* 2008; 16: 9-12
- 4).Karosaki H, Sharma RK., Alok T, The systematic generation of monodentate bridging ethanoates in dinuclear nickel(II) complexes of asymmetric compartmental ligands. *J. Chem. Soc. Dalton trans.* 2002; 441-448
- 5).Desai B, Sureja D, Nalpara Y, Synthesis and QSAR Studies of 4-Substituted phenyl-2,6-dimethyl-3,5-bis-N-(substituted phenyl) carbamoyl-1,4-dihydro pyridines as potential anti tubercular agents. *Bioorg. Med. Chem.* 2001; 9:1993-1998

agents. *Bioorg. Med. Chem.* 2001; 9:1993-1998

6).Trost BM, Hachiya I, A Convergent Synthesis of (+) Parviflorin, (+)-Squamocin K, and (+)-5S-Hydroxyparviflorin. *J. Am. Chem. Soc.*, 1998; 120:1104

7).Desai B, Sureja D, Nalpara Y, Synthesis and QSAR Studies of 4-Substituted phenyl-2,6-dimethyl-3, 5-bis-N-(substituted phenyl) carbamoyl-1,4-dihydropyridines as potential antitubercular agents. *Bioorg. Med. Chem.* 2001; 9:1993-1998

8).Safak C, Simsek R, **Fused 1,4-Dihydropyridines as Potential Calcium Modulatory Compounds**, *Mini. Rev. Med. Chem.* 2006; 67:747-755

9).Tasaka H S, Ohmori H, Synthesis and structure-activity analysis of novel dihydropyridine derivatives to overcome multidrug resistance. *Bioorg. Med. Chem. Lett.* 2001; 11:275-277

10).Collision SR, Gelbrick T, Hurhtouse MB, Novel ferrocene receptors for barbiturates and ureas. *Chem. Comm.* 2001; 6:555-556

11).Huc I, Krishche MJ, Funeria P, Dynamic Combinatorial Chemistry: Substrate H-Bonding Directed Assembly of Receptors Based on Bipyridine - Metal Complexes. *Eur. J. Inorg. Chem.* 1999; 9:1415-1420

12).Epperson JD, Ming LJ, Baker GR, Paramagnetic Cobalt(II) as an NMR Probe for Dendrimer Structure: Mobility and Cooperativity of Dendritic Arms. *J. Am. chem. Soc.* 2001; 123: 8583-8592

13).Huc I, Krishche MJ, Funeria P, Dynamic Combinatorial Chemistry: Substrate H-Bonding Directed Assembly of Receptors Based on Bipyridine - Metal Complexes. *Eur. J. Inorg. Chem.* 1999; 9: 1415-1420

14).Cooper K, Dihydropyridine anti-allergic and anti-inflammatory agents. *Eur. Patent* 294074, 1998

15).Kumar PA, Synthesis and spectral investigation of manganese(II), cadmium(II) and oxovanadium(IV) complexes with 2,6-diacetylpyridine bis(2-aminobenzoylhydrazone): Crystal structure of manganese(II) and cadmium(II) complexes. *Ind. J. Chem.* 1998; 37(A),460

16).Gupta MD, Nag S, Das G, N,N'-Bis(aryl)pyridine-2,6-dicarboxamide complexes of ruthenium: Synthesis, structure and redox properties. *Polyhedron*, 2008; 27:139-150

17).Dana S, Marlin M, Olmstead M, Bis-(N,N'-bis[2-(2-pyridyl) methyl] pyridine-2,6-dicarboxamido)-dicopper(II): spontaneous formation of a short double stranded helicate *Inorg. Chim Acta.* 2002; 323: 1-4

18).Zang J, Liu Q, Dunan C, Shao Y, Structural evidence for the facile chelate-ring opening reactions

of novel platinum(II)–pyridine carboxamide complexes, *J. Chem. Soc. Dalton trans.* 2002, 591

**19).** Kalagouda B, Gudsai A, Siddpa A, Patil SA, Synthesis and spectral studies of Cu(II), Ni(II), Co(II), Mn(II), Zn(II) and Cd(II) complexes of a new macrocyclic ligand *n,n'*-bis(2-benzothiazolyl)-2,6-pyridinedicarboxamide. *J. Serb. Chem. Soc.* 2006, 71:(5), 529-542

**20).** Jain SL, Bhattacharya P, Patil SA, New pyridine carboxamide ligands and their complexation to copper(II). X-Ray crystal structures of mono-, di, tri- and tetranuclear copper complexes. *J. Chem. Soc. Dalton Trans.* 2004; 4:862-871

**21).** Smail KZ, Dissouky AE, Shehada AZ, Spectroscopic and magnetic studies on some copper (II) complexes of antipyrine Schiff base derivatives. *Polyhedron*, 1997; 16: 2909- 2916

